Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 293 KMean $\sigma(C-C) = 0.005 \text{ Å}$ R factor = 0.064 wR factor = 0.200 Data-to-parameter ratio = 13.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

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In the title compound, $C_{28}H_{27}NO_5$, the pyran ring in the chroman-4-one moiety adopts a sofa conformation and the pyrrolidine ring adopts an envelope conformation. The chromanone moiety makes a dihedral angle of 79.1 (1)° with the mean plane passing through the pyrrolidine ring. In the crystal structure, the molecules in the crystal lattice are

stabilized by $C-H \cdots O$ intermolecular interactions.

Spiro[5-carbomethoxy-2-(4-methoxyphenyl)-4-

(4-methylphenyl)pyrrolidine-3,3'-chroman-4'-one]

Received 22 March 2002 Accepted 17 May 2002 Online 24 May 2002

Comment

Highly substituted pyrrolidines have attracted much interest in the past few years, since they constitute the main structural element of many alkaloids and pharmacologically active compounds (Waldmann, 1995). Many 4-chromanone derivatives are used as versatile intermediates in the synthesis of natural products such as hematoxylin, brazillin, ripariochromene, clausenin, calanolide A and inophyllum B (Ellis *et al.*, 1977; Chenera *et al.*, 1993). It has been suggested that they have significant activity against human immunodeficiency virus type I (HIV-1) (Hussain & Amir, 1986).



The title compound, (I), differs from a closely related compound (Nissa *et al.*, 2001) by having an extra methyl group at the *para* position of the phenyl ring (C22–C27). The total puckering amplitude (Cremer & Pople, 1975) of the pyran ring in the chroman-4-one moiety ($Q_T = 0.5093$) and the values of the lowest displacement asymmetry parameters (Nardelli, 1983*a*) [Δ_2 (C3–C2) = 0.102 (1) and Δ_s (C3) = 0.035 (2)] are indicative of a sofa conformation. Calculation shows that atom C3 is at the apex and deviates by 0.681 (3) Å from the mean plane passing through the other atoms (C1, C2, O2, C9 and C8) of the ring. The puckering parameters of the fivemembered pyrrolidine ring are $q_2 = 0.4174$ and $\Phi_2 = 29.32$, and the value of the lowest displacement asymmetry parameter Δ_s (C18) = 0.042 (2), indicative of an envelope conformation.

The chromanone moiety makes a dihedral angle of 79.1 $(1)^{\circ}$ with the mean plane passing through the pyrrolidine ring. The



Figure 1

ZORTEP (Zsolnai, 1997) diagram of the structure showing the atomlabelling scheme. Displacement ellipsoids are drawn at the 30% probability level.

dihedral angle between the planes of the fused aromatic ring and the methoxyphenyl ring is $46.1 (1)^{\circ}$. The dihedral angle [89.1 (1)°] shows that the fused aromatic ring and the *p*-tolyl ring are perpendicular to each other. The methoxyphenyl group is oriented at an angle of 74.5 $(1)^{\circ}$ to the pyrrolidine ring. The phenyl rings make a dihedral angle of $66.9 (1)^{\circ}$.

The twisting of the *p*-tolyl ring can be described by the torsion angle C2-C21-C22-C27 of $83.8 (4)^{\circ}$. The torsion angle C17-O3-C14-C13 of -178.8 (5)° shows that the methoxy substituent is in an antiperiplanar conformation. The molecules in the crystal lattice are stabilized by intermolecular C-H···O interactions in addition to van der Waals contacts (Fig. 2). As a result of the additional methyl group, molecules in the crystal lattice pack in a different manner, compared to that of the previously reported related structure (Nissa et al., 2001). Excluding the methyl group C28, superposition of the non-H atoms of the compound with the reported compound (Nissa et al., 2001) [using BIOSYM INSIGHT-II (MSI, 1995)] shows that the r.m.s deviation for all the non-H atoms in the entire molecule is 2.2 Å, while the r.m.s deviation for the non-H atoms of the chromanone moiety is 0.3 Å. The r.m.s deviation for the non-H atoms of both compounds comprising the pyrrolidine ring is 0.3 Å.

Experimental

An efficient synthesis of a series of novel spiropyrrolidine derivatives has been achieved via the [3+2]-cycloaddition reaction between



Figure 2 Stereoview of the molecular packing of the title compound, viewed down the c axis.

N-metalated azomethine ylides and (E)-3-arylidene-4-chromanones (Subramaniyan & Raghunathan, 2001). The following is the general procedure for the cycloaddition reaction between 3-arylidene-4chromanones and imines in the presence of silver acetate as catalyst: to a solution of benzylideneglycine ester (1 mmol) in dry acetonitrile (10 ml), triethylamine (1 mmol), benzylidenechromanone (1 mmol) and then AgOAc (0.15 equivalents) were added. After completion of the reaction as determined by TLC, the reaction mixture was filtered through a celite pad, washed with a saturated aqueous solution of NH_4Cl and then extracted with CH_2Cl_2 (2 \times 20 ml). The combined organic layers were washed with brine, dried (MgSO₄), filtered and the solvent evaporated in vacuo. The residue was purified by column chromatography on silica gel (100-200 mesh) with petroleum etherethyl acetate (4:1) to afford the cycloadduct, which was crystallized from EtOH (0.31 g, 68%; m.p.: 425–426 K).

Crystal data

$C_{28}H_{27}NO_5$	$D_x = 1.289 \text{ Mg m}^{-3}$
$M_r = 457.51$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters fro
a = 12.4779 (6) Å	reflections
b = 18.0879 (19) Å	$\theta = 19.7 - 28.2^{\circ}$
c = 10.4580 (19) Å	$\mu = 0.09 \text{ mm}^{-1}$
$\beta = 92.782 \ (9)^{\circ}$	T = 293 (2) K
$V = 2357.6 (5) \text{ Å}^3$	Prism, colourless
Z = 4	$0.32 \times 0.25 \times 0.18$
Data collection	
Enraf-Nonius CAD-4	$\theta_{\rm max} = 25.0^{\circ}$
diffractometer	$h = 0 \rightarrow 14$
$\omega/2\theta$ scans	$k = -21 \rightarrow 0$
Absorption correction: none	$l = -12 \rightarrow 12$
4342 measured reflections	3 standard reflectio
4144 independent reflections	every 100 reflect
2306 reflections with $I > 2\sigma(I)$	frequency: 120 m
$R_{\rm int} = 0.079$	intensity decay:
Refinement	
	2 2

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.064$ $wR(F^2) = 0.200$ S=1.024144 reflections 310 parameters H-atom parameters constrained

radiation rameters from 25 ctions 7-28.2 09 mm^{-1} 3(2)K colourless $0.25 \times 0.18 \text{ mm}$ 25.0°

 $\rightarrow 14$ $21 \rightarrow 0$ $2 \rightarrow 12$ lard reflections y 100 reflections uency: 120 min nsity decay: 1%

 $w = 1/[\sigma^2(F_o^2) + (0.1092P)^2]$ + 0.3704P] where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{\rm max} = 0.005$ $\Delta \rho_{\rm max} = 0.34 \ {\rm e} \ {\rm \AA}^{-3}$ $\Delta \rho_{\rm min} = -0.41 \text{ e } \text{\AA}^{-3}$

Table 1Hydrogen-bonding geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdots A$	$D \cdots A$	$D - H \cdots A$
$\overline{\text{C3}-\text{H3}A\cdots\text{O4}^{i}}$	0.97	2.62	3.591 (4)	175
Symmetry code: (i) x	$-\frac{1}{2} - v, \frac{1}{2} + z$			

All H atoms were fixed geometrically and refined isotropically using a riding model.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD*4 (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ZORTEP* (Zsolnai, 1997) and *BIOSYM INSIGHT*-II (MSI, 1995); software used to prepare material for publication: *SHELXL97* and *PARST* (Nardelli, 1983*b*, 1995).

DV and VR thank the Department of Science and Technology, India. MN thanks the CSIR for SRF.

References

- Chenera, B., West, M. L., Finkelstein, J. A. & Dreyer, G. B. (1993). J. Org. Chem. 58, 5605–5606.
- Cremer, D. & Pople, J. A. (1975). J. Am. Chem. Soc. 97, 1354-1358.
- Ellis, G. P., Lockhart, I. M., Meeder-Nycz, D. & Schweizer, E. E. (1977). *Chromenes, Chromanones and Chromones*, edited by G. P. Ellis. New York: John Wiley and Sons Inc.
- Enraf-Nonius (1994). CAD-4 EXPRESS. Version 5.1/1.2. Enraf-Nonius, Delft, The Netherlands.
- Harms, K. & Wocadlo, S. (1995). XCAD4. University of Marburg, Germany.
- Hussain, M. I. & Amir, M. (1986). J. Indian Chem. Soc. 63, 317-320.
- MSI (1995). *BIOSYM INSIGHT*-II. Release 95.0. Molecular Simulations Inc., San Diego, USA.
- Nardelli, M. (1983a). Acta Cryst. C39, 1141-1142.
- Nardelli, M. (1983b). Comput. Chem. 7, 95-97.
- Nardelli, M. (1995). J. Appl. Cryst. 28, 659.
- Nissa, M. N., Rajakannan, V., Kim, M. J. & Velmurugan, D. (2001). Acta Cryst, E57, 01230-01232.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Subramaniyan, G. & Raghunathan, R. (2001). Personal communication.
- Waldmann, H. (1995). Synlett, pp. 133-141.
- Zsolnai, L. (1997). ZORTEP. University of Heidelberg, Germany.